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A. V. Levkovskii; A.Yu. Aksinenko; A. N. Pushin; V. B. Sokolov

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Fluorinated 1-Methylaminoalkylphosphonates. Interaction with Ammonia and Methylamine

A.V. LEVKOVSKII, A.YU. AKSINENKO, A.N. PUSHIN and V.B. SOKOLOV

Institute of Physiologically Active Substances of the Russian Academy of Sciences, Chernogoloyka, Moscow region 142432, Russia

Previously we reported that fluorinated aminophosphonates $\underline{1a}$ react with NH₃ to form heterocyclic salts $\underline{2}$ [1,2]. As was found, $\underline{2}$ eliminate NH₃ under heating to give neutral 1,4,2-diazaphospholines $\underline{3}$. The reverse reaction of $\underline{3}$ with NH₃, any amines, and alkalis led to the same aniones $\underline{2}$. Salts $\underline{2}$ and diazaphospholines $\underline{3}$ are hydrolyzing by moisture of ammonia or solvents into betaines $\underline{4}$.

Aminophosphonates $\underline{1b}$ (H at C^1) react with NH₃ to form, on the whole, linear guanidinophosphonates $\underline{4}$ and $\underline{5}$, while $\underline{2}$ were detected as traces.

Reaction of $\underline{1}$ with an excess of CH₃NH₂ resulted in betaines $\underline{7}$ [3] and $\underline{8}$ obviously through the same stages.

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